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2.10 What role does the molecular interaction play in a solution of alcohol and water? Sol. Alcohol and water both have strong tendency to form intermolecular hydrogen bonding. On mixing the two, a solution is formed as a

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result of formation of H-bonds between alcohol and H_2O molecules but these interactions are weaker and less extensive than those in pure H_2O .

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1.1 - General Characteristics of Solid
State Ex 1.2 - Amorphous and Crystalline
Solids

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