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2) It does not require constant attention. 3) Equilibrium is quickly attained, hence it is not so time-consuming. 4) A small quantity of solution is needed. 5) Air need not be eliminated.

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1. All matter is composed of atoms 2. Atoms of the same element have the same chemical properties and atoms of different elements have different chemical properties 3. Atoms combine in definite ratios to form compounds 4. In a Chemical reaction, atoms are rearranged to form new compounds

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1 AP Chemistry A. Allan Chapter 1 Notes - Chemical Foundations  
1.1 Chemistry: An Overview A. Reaction of hydrogen and oxygen  
1. Two molecules of hydrogen react with one molecule of oxygen to form two molecules of water  $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$   
2. Decomposition of water  $2\text{H}_2\text{O} \rightarrow 2\text{H}_2 + \text{O}_2$   
B. Problem Solving in Chemistry (and life) 1.

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CHAPTER 1: FUNDAMENTALS OF CHEMISTRY The first chapter is  
all about the basic of Chemistry from the definition of Chemistry  
to its main branches, from empirical formula to formula mass,  
from Chemical species to Avogadro's number.

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