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Study Guides

STUDY GUIDE In your textbook, read about organic chemistry, hydrocarbons, and straight-chain alkanes. Use each of the terms below just once to complete the passage. homologous series straight-chain alkanes If all of the carbon atoms are linked by single covalent bonds and there are no branches, the compounds

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are called (1) examples.

KMBT 654-20150625115832

ChemFacts. • The primary source of hydrocarbons is petroleum. • About 75 million barrels of petroleum are pumped out of the Earth each day. •

Hydrocarbons are used as fuels and are the raw materials for products such as

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plastics, synthetic fibers, solvents, and industrial chemicals.

Chapter 21: Hydrocarbons

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Hydrocarbons Answer

Study Guide Key Concepts 22.1

Hydrocarbons • Because carbon has four valence electrons, carbon atoms always form four covalent bonds. • The carbon atoms in an alkane can be arranged in a straight chain or in a chain that has branches. • Molecules of hydrocarbons, such as alkanes, are nonpolar

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molecules. 22.2 Unsaturated Hydrocarbons

CHAPTER 22 Study Guide - Quia

Chapter 22 Hydrocarbons Study Guide Answers Hydrocarbons are organic molecules that only contain two elements: carbon and hydrogen. Common hydrocarbons include

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methane, propane, and butane.
Hydrocarbons are the primary constituents of fossil...

Hydrocarbon Study Guide Answers

Hydrocarbons; Quiz: Hydrocarbons;
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Chemistry is the study of matter: its composition, properties, and reactivity. This material roughly covers a first-year high school or college course, and a good understanding of algebra is helpful. Did you know that everything is made out of chemicals? Chemistry is the study of matter: its composition, properties, and reactivity.

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Chemistry | Science | Khan Academy

Organic Chemistry and Hydrocarbons

(pages 693-694) 1. ... 238 Guided

Reading and Study Workbook CHAPTER

22, Hydrocarbon Compounds (continued)

14. Circle the letter of each condensed structural formula for pentane. a. C_5H_{12} b. $CH_3CH_2CH_2CH_2CH_3$ c. CH_3

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(CH₂)₃CH₃ d. C — C — C — C — C 15.

SECTION 22.1

HYDROCARBONS(pages 693-701)

Chapter 17 - An Introduction to Organic Chemistry, Biochemistry, and Synthetic Polymers 271 Key Ideas Answers 8.

Hydrocarbons (compounds composed of carbon and hydrogen) in which all of the

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carbon-carbon bonds are single bonds are called alkanes. 10. When a(n) small section of an organic molecule is largely responsible for the molecule's

Chapter 17 An Introduction to Organic Chemistry ...

Benzene, C_6H_6 , is the simplest member of a large family of

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hydrocarbons, called aromatic hydrocarbons. These compounds contain ring structures and exhibit bonding that must be described using the resonance hybrid concept of valence bond theory or the delocalization concept of molecular orbital theory.

21.7: Aromatic Hydrocarbons -

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at the

Hazmat Chemistry Study Guide (second edition)

2 A Self-Study Guide to the Principles of Organic Chemistry 6.5 Electron Configuration of the Elements Commonly Involved in Organic Compounds Figure 1.1 Hydrocarbon,

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methane and its derivatives The Periodic Table contains more than one hundred elements; however, organic chemistry deals mostly with only certain elements.

A SELF-STUDY GUIDE TO THE PRINCIPLES OF ORGANIC CHEMISTRY

Chemistry: Matter and Change Study

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Guide 16 21 Hydrocarbons Section 21.1
Introduction to Hydrocarbons In your
textbook, read about natural sources of
hydrocarbons and rating gasolines. In
the space at the left, write the word or
phrase in parentheses that correctly
completes the statement. _____ 1.

Organic Chemistry Worksheet -

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Homestead

Study Guide Key Concepts 221

Hydrocarbons • Because carbon has four valence electrons, carbon atoms always form four covalent bonds • The carbon atoms in an alkane can be arranged in a straight chain or in a chain that has branches • Molecules of hydrocarbons, such as alkanes, are nonpolar molecules

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222 Unsaturated

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