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Another factor entering into the process of solution formation is the increase (or occasionally, the decrease) in the entropy — that is, the degree to which thermal energy is dispersed or “diluted”. Explain this in your own terms. Explain how the adage “like dissolves like” reflects the effects mentioned above.

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The first law of thermodynamics - problems and solutions. 1. 3000 J of heat is added to a system and 2500 J of work is done by the system. What is the change in internal energy of the system? Known : Heat (Q) = +3000 Joule. Work (W) = +2500 Joule . Wanted: the change in internal energy of the system Solution :

The first law of thermodynamics - problems and solutions ...

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